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Applicant: Colgate-Palmolive Company  
300 Park Avenue  
New York, N.Y. 10022-7499(US)

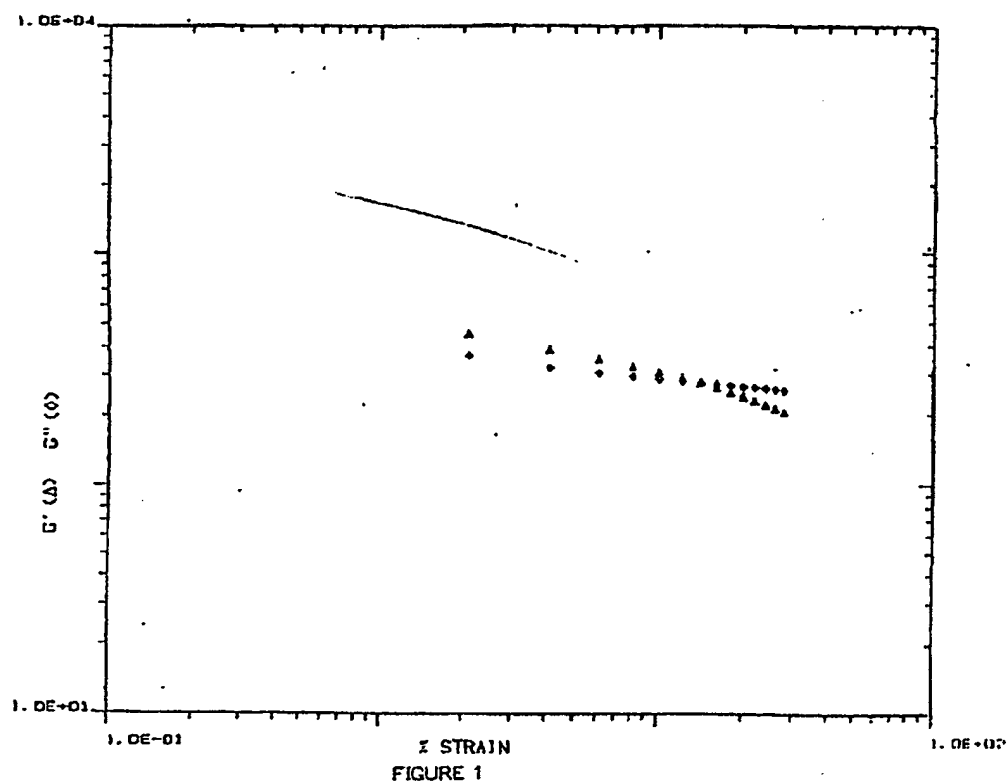
Inventor: Dixit, Nagaraj S.  
3 Titus Lane  
Plainsboro, New Jersey(US)  
Inventor: Kenkare, Divaker  
Mountaiview Road  
Asbury, New Jersey(US)  
Inventor: Shevade, Makaran  
65 Willow Court  
Hamilton, N.J.(US)  
Inventor: Phillips, Patricia  
2 Frewert Lanet  
Sommerville, N.J.(US)

Representative: UEXKÜLL & STOLBERG  
Patentanwälte  
Beselerstrasse 4  
W-2000 Hamburg 52(DE)

Linear viscoelastic aqueous liquid detergent compositions, especially for automatic dishwashers, or improved high temperature stability.

Automatic dishwasher detergent composition is formulated as a linear viscoelastic, pseudoplastic, gel-like aqueous product of exceptionally good physical stability, low bottle residue, low cup leakage, and improved cleaning performance. Linear viscoelasticity and pseudoplastic behavior is attributed by incorporation of cross-linked high molecular weight polycarboxylate acid type thickener. Potassium to sodium weight ratios of at least 1/1 minimize amount of undissolved solid particles to further contribute to stability and pourability. Stearic acid or other higher fatty acid mixture or salt further improves physical stability. Use of potassium triphosphosphate in place of tetrapotassium pyrophosphate reduces formation of crystal growth when the product is subjected to elevated temperatures, such as 100° F.

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Field of Invention

The invention relates to aqueous linear viscoelastic (gel-like) liquid compositions which are especially useful as automatic dishwasher detergent compositions.

Background of the Invention

Liquid automatic dishwasher detergent compositions, both aqueous and nonaqueous, have recently received much attention, and the aqueous products have achieved commercial popularity.

10 The acceptance and popularity of the liquid formulations as compared to the more conventional powder products stems from the convenience and performance of the liquid products. However, even the best of the currently available liquid formulations still suffer from two major problems, product phase instability and bottle residue, and to some extent cup leakage from the dispenser cup of the automatic dishwashing machine.

15 Representative patent art in this area includes Rek, U.S. Patent 4,556,504; Bush, et al., U.S. Patent 4,226,736; Ulrich, U.S. Patent 4,431,559; Sabatelli, U.S. Patent 4,147,650; Paucot, U.S. Patent 4,079,015; Leikhem, U.S. Patent 4,116,849; Milora, U.S. Patent 4,521,332; Jones, U.S. Patent 4,597,889; Heile, U.S. Patent 4,512,908; Laitern, U.S. Patent 4,753,748; Sabatelli, U.S. Patent 3,579,455; Hynam, U.S. Patent 3,684,722. Other patents relating to thickened aqueous detergent compositions include U.S. Patent 20 3,985,668; U.K. Patent Applications GB 2,116,199A and GB 240,450A; U.S. Patent 4,511,487; U.S. Patent 4,752,409 (Drapier, et al.); U.S. Patent 4,801,395 (Drapier, et al.); U.S. Patent 4,836,948 (Dixit); U.S. Patent 4,889,653 (Ahmed, et al.). Commonly assigned co-pending patents include, for example, Serial No. 427,912 filed October 24, 1989; Serial No. 924,385, filed October 29, 1986; Serial No. 323,138, filed March 13, 1989; Serial No. 328,716, filed March 27, 1989; Serial No. 323,137, filed March 13, 1989; Serial No. 323,134, filed 25 March 13, 1989.

The solubilizing effect of potassium salts on sodium tripolyphosphate in aqueous detergent compositions is described in U.S. Patent 3,720,621 (Smeets). This patent describes homogeneous liquid compositions containing 14 to 35 percent sodium tripolyphosphate, 0.1 to 50 percent of potassium and/or ammonium salt of an inorganic or organic acid, water, and optional surfactants, solubilizing agent, organic 30 sequestering agent, per-compound, and other adjuvants. Potassium pyrophosphate (TKPP) is disclosed as the preferred potassium salt for its sequestering and solubilizing activity and also because "it effectively stabilizes the tripolyphosphate agent hydrolysis in aqueous solution into sodium pyrophosphate and sodium orthophosphate." The patent also states that, "[t]he content [of TKPP] should not ... be too great so as to undergo a double-decomposition reaction with TPP to form sodium pyrophosphate, which is well known, 35 has a very low solubility in water and, therefore, tends to separate out as a precipitate. The applicants investigations have shown that the optimum proportion of potassium pyrophosphate incorporated in the liquid detergent compositions ... is between 5 and 25 percent based on the TPP present in the compositions."

40 U.S. Patent 4,836,948 discloses a viscoelastic gel detergent composition characterized by its viscosity under low and high shear conditions, pH, and steady state viscoelastic deformation compliance. The composition requires the presence of a polycarboxylate polymeric thickener, preferably a cross-linked polyacrylic acid. The compositions of this patent also, however, require a trivalent metal containing material, especially an aluminum containing material such as alumina. The compositions may further include a structuring chelant which may be a salt of carbonate, pyrophosphate or mixture thereof, and preferably the 45 potassium salts.

The recently issued U.S. Patent 4,859,358 discloses the incorporation of metal salts of long chain hydroxy fatty acids, as anti-tarnishing agents, in thickened aqueous automatic dishwashing detergent compositions. The thickeners for these compositions may be a high molecular weight polycarboxylate polymer, such as those sold under the Carbopol trade name, and specific 600 and 900 series resins are 50 mentioned. It is also disclosed that the compositions may include entrained gas, e.g. air, bubbles to further ensure stability. Amounts of air in the range of from about 1% to about 20%, preferably from about 5 to 15% by volume will lower the specific gravity of the overall composition to within from about 5% more than to about 10% less than, preferably from 1% more than to 5% less than, the specific gravity of the aqueous phase. In Example III of this patent, the specific gravity of the composition was stated as 1.32 g/cm<sup>3</sup>. The 55 compositions are not described as being linear viscoelastic and as exemplified do not include any potassium salts.

While the compositions disclosed in our prior application Serial No. 353,712 provided a satisfactory solution to the problems of phase instability and bottle residue, as well as cup leakage, it has now been

found that under some storage/handling conditions and/or processing conditions, additional improvements would be desirable. Specifically, if the viscoelastic composition is subjected to repeated heating and cooling cycles, growth of crystals and product thinning and/or precipitate formation has been observed. Chemical analysis of the precipitated crystals has shown that these crystals are comprised predominantly of sodium pyrophosphate. In addition, it appears that the crystals tend to become entangled with the polymeric thickener which tendency is presumed to account for the thinning out or aqueous phase separation which has been observed in conjunction with crystal formation and precipitation.

Another problem which has been encountered in the preparation of the linear viscoelastic gel formulations of our prior application involved the manner by which the preferred Carbopol thickener is added to water. Specifically, if in forming the slurry of the Carbopol type thickener, the water temperature is maintained at ambient temperature conditions, e.g. 60°-80°F, the polymer tends to be only incompletely hydrated, i.e. results in formation of fish eyes.

As a result of these problems, the finished product not only may be perceived as unaesthetic but, more importantly, the product viscosity is often lowered which in turn may lead to cup leakage and corresponding degradation in cleaning performance.

It has also been discovered that there is an important relationship between the pH of the composition at each stage during processing and product stability, including both phase separation and loss of viscosity.

The present invention provides ingredient and processing modifications to our previously developed compositions which help to alleviate and overcome the above problems.

#### Brief Description of the Drawings

Figures 1-6 are rheograms, plotting elastic modulus  $G'$  and viscous modulus  $G''$  as a function of applied strain, for the compositions of Example 1, Formulations A, B, C, D, E and F, respectively.

#### Summary of the Invention

According to the present invention there is provided an improved aqueous liquid automatic dishwasher detergent composition. The composition is characterized not only by its linear viscoelastic behavior, substantially indefinite stability against phase separation or settling of dissolved or suspended particles under high and low temperature conditions, low levels of bottle residue, relatively high bulk density, and substantial absence of loosely bound water, but also by its overall product consistency from batch to batch and run to run and over a wide range of storage and aging conditions, including superior aesthetics, freedom from fish-eyes, absence of crystal formation and growth, and resistance to cup leakage.

The present invention was accomplished based, in part, on the discovery that by replacing part or all of the tetrapotassium pyrophosphate (TKPP) with potassium tripolyphosphate (KTPP), crystal growth can be inhibited and product consistency can be improved and the linear viscoelastic properties can be extended beyond that previously achieved. Further improvements are obtained by utilizing more highly cross-linked and branched, temperature stable and bleach and salt compatible polymeric thickening agents. Furthermore, by controlling the conditions of mixing the ingredients of the compositions, and particularly the pH and temperature of the aqueous phase used to form the aqueous solution of the polymeric thickener, formation of fish-eyes and elimination of viscosity reductions with time can be achieved.

Accordingly, in one aspect the present invention provides an improved linear viscoelastic aqueous liquid automatic dishwasher detergent composition comprising water, up to about 2% by weight of long chain fatty acid or salt thereof, from about 0.1 to 5% by weight of low-foaming chlorine bleach stable, water dispersible automatic dishwasher non-soap organic detergent, from about 10 to 35% by weight of alkali metal detergent builder salt, at least 50% by weight of which is potassium tripolyphosphate and at least about 5% by weight of which is sodium tripolyphosphate, from about 3 to 20% by weight of a chlorine bleach compound, and a cross-linked polycarboxylate thickening agent having a molecular weight of at least about 500,000, wherein the aqueous phase includes both sodium and potassium ions at a K/Na weight ratio of from about 1/1 to about 45/1. The compositions preferably have a bulk density of from about 1.28 g/cm<sup>3</sup> to about 1.42 g/cm<sup>3</sup>.

In a preferred embodiment, the linear viscoelastic aqueous liquid automatic dishwasher detergent comprises, approximately, by weight,

(a)

- (i) 8 to 25% potassium tripolyphosphate detergent builder,
- (ii) 2 to 10% sodium tripolyphosphate detergent builder, at an (i)/(ii) weight ratio of from about 1.4/1 to 10/1;

(b) 5 to 15% alkali metal silicate;

- (c) 1 to 6% alkali metal hydroxide;
- (d) 0.1 to 3% chlorine bleach stable, water-dispersible organic detergent active material;
- (e) 0.05 to 1.5% chlorine bleach stable foam depressant;
- (f) chlorine bleach compound in an amount to provide about 0.2 to 4% of available chlorine;
- (g) 0.4 to 1.5% hydrophilic cross-linked water insoluble polycarboxylate thickening agent having a molecular weight of from 800,000 to 4,000,000 to provide said linear viscoelastic property;
- (h) 0.08 to 0.4% of long chain fatty acid or a metal salt of a long chain fatty acid as a physical stabilizer to increase the physical stability of the composition;
- (i) 0 to 10% of a non-cross-linked polyacrylic acid having a molecular weight in the range of from about 800 to 200,000; and
- (j) water;

and wherein in the entire composition, the ratio, by weight, of potassium and sodium is from about 1.05/1 to 3/1.

In another aspect of the invention, a method for preparing the aqueous linear viscoelastic composition is provided. According to this aspect, the method comprises the steps of

- I.
  - (a) fully hydrating the cross-linked polycarboxylate thickener by slowly adding the thickener to heated water while moderately agitating the mixture,
  - (b) slowly adding a neutralizing amount of caustic soda to the mixture from (a) while continuing agitation to obtain a dispersion of the neutralized thickener;
- II.
  - (c) forming an aqueous mixture of surface active agents;
  - (d) heating the mixture in (c) to a temperature higher than that of the heated water in (a) and mixing until a homogeneous smooth premix is obtained;
- III.
  - (e) uniformly mixing alkali metal builder salts with the dispersion (b),
  - (f) uniformly mixing the heated premix (d) with the mixture (e),
  - (g) cooling the mixture (f) to a temperature above the temperature of the heated water in step (a), and
  - (h) adding bleach to the mixture (g).

In a preferred embodiment of the invention process, the pH of the aqueous slurry of the cross-linked polycarboxylate thickener after the neutralization in step (b) and in each succeeding step is maintained at a value of at least 11.

#### Detailed Description and Preferred Embodiments

The compositions of this invention are thickened aqueous liquids containing various cleansing active ingredients, detergent builder salts and other detergent adjuvants, structuring and thickening agents and stabilizing components, although some ingredients may serve more than one of these functions.

The advantageous characteristics of the compositions of this invention, include physical stability, as manifested by little or no phase separation, solid settling or viscosity changes over time or resulting from exposure to low or high temperatures, low bottle residue, low cup leakage, high cleaning performance, e.g. low spotting and filming, dirt residue removal, and consistency in product characteristics and performance, and superior aesthetics. These characteristics are believed to be attributed to several interrelated factors such as low solids, i.e. undissolved particulate content, product density and linear viscoelastic rheology. These factors are, in turn, dependent on several critical compositional components and processing conditions of the formulations, namely, (1) the inclusion of a thickening effective amount of a temperature, bleach and salt stable cross-linked polymeric thickening agent having high water absorption capacity, exemplified by cross-linked polyacrylic acid, (2) inclusion of a physical stabilizing amount of a long chain fatty acid or salt thereof, and (3) potassium ion to sodium ion weight ratio K/Na in the range of from about 1:1 to 45:1, especially from 1:1 to 3:1, and preferably also (4)-(6): (4) a product bulk density of at least about 1.28 g/cc, especially at least 1.32 g/cc, (5) hydrating the cross-linked polymeric thickener with heated water, and (6) maintaining the pH of the neutralized polymeric thickener at a pH of at least 11, preferably at least 11.5.

In particular, the linear viscoelastic aqueous liquid automatic dishwasher detergent compositions of this invention will, at least in the preferred embodiments, satisfy each of the following stability criteria over the aging temperature-time schedule shown by the following Table A:

**TABLE A**

Aging Temperature (°F)	Duration (Weeks)	
	Minimum	Preferred
140	≥ 1	≥ 2
120	≥ 6	≥ 8
100	≥ 13	≥ 16
Ambient	≥ 24	≥ 24

More specifically, the compositions are considered stable if each of the following stability criteria is satisfied for at least the minimum number of weeks for each aging temperature shown in Table A:

- no visible phase separation (i.e. no solid/liquid separation)
- no significant change (e.g. less than 10%) in viscosities, yield stress or other dynamic-mechanical properties
- no crystal growth under repeated heating-cooling cycles over a temperature range of at least 7° F to 140° F
- no decolorization or significant color change.

In addition to the above stability criteria, the compositions of this invention are further characterized by their low bottle residue. Specifically, for the preferred cross-linked polyacrylic acid thickened compositions of this invention, bottle residues, under the usual use conditions, will be no more than about 6 to 8%, preferably no more than about 4 to 5%, of the original bottle contents, on a weight basis.

The polymeric thickening agents contribute to the linear viscoelastic rheology of the invention compositions. As used herein, "linear viscoelastic" or "linear viscoelasticity" means that the elastic (storage) moduli ( $G'$ ) and the viscous (loss) moduli ( $G''$ ) are both substantially independent of strain, at least in an applied strain range of from 0-50%, and preferably over an applied strain range of from 0 to 80%. More specifically, a composition is considered to be linear viscoelastic for purposes of this invention, if over the strain range of 0-50% the elastic moduli  $G'$  has a minimum value of 100 dynes/sq.cm., preferably at least 250 dynes/sq.cm., and varies less than about 500 dynes/sq.cm., preferably less than 300 dynes/sq.cm., especially preferably less than 100 dynes/sq.cm. Preferably, the minimum value of  $G'$  and maximum variation of  $G'$  applies over the strain range of 0 to 80%. Typically, the variation in loss moduli  $G''$  will be less than that of  $G'$ . As a further characteristic of the preferred linear viscoelastic compositions the ratio of  $G''/G'$  ( $\tan \delta$ ) is less than 1, preferably less than 0.8, but more than 0.05, preferably more than 0.2, at least over the strain range of 0 to 50%, and preferably over the strain range of 0 to 80%. It should be noted in this regard that % strain is shear strain x100.

By way of further explanation, the elastic (storage) modulus  $G'$  is a measure of the energy stored and retrieved when a strain is applied to the composition while viscous (loss) modulus  $G''$  is a measure of the amount of energy dissipated as heat when strain is applied. Therefore, a value of  $\tan \delta$ ,

$$0.05 \leq \tan \delta < 1,$$

preferably

$$0.2 \leq \tan \delta \leq 0.8$$

means that the compositions will retain sufficient energy when a stress or strain is applied, at least over the extent expected to be encountered for products of this type, for example, when poured from or shaken in the bottle, or stored in the dishwasher detergent dispenser cup of an automatic dishwashing machine, to return to its previous condition when the stress or strain is removed. The compositions with  $\tan \delta$  values in these ranges, therefore, will also have a high cohesive property, namely, when a shear or strain is applied to a portion of the composition to cause it to flow, the surrounding portions will follow. As a result of this cohesiveness of the linear viscoelastic characteristic, the compositions will readily flow uniformly and homogeneously from a bottle when the bottle is tilted, thereby contributing to the physical (phase) stability of the formulation and the low bottle residue (low product loss in the bottle) which characterizes the invention compositions. The linear viscoelastic property also contributes to improved physical stability against phase separation of any undissolved suspended particles by providing a resistance to movement of the particles due to the strain exerted by a particle on the surrounding fluid medium. Linear viscoelasticity

also contributes to the elimination of dripping of the contents when the product is poured from a bottle and hence reduction of formation of drops around the container mouth at the conclusion of pouring the product from a container.

Also contributing to the physical stability and low bottle residue of the invention compositions is the high potassium to sodium ion ratios in the range of 1:1 to 45:1, preferably 1:1 to 4:1, especially preferably from 1.05:1 to 3:1, for example 1.1:1, 1.2:1, 1.5:1, 2:1, or 2.5:1. At these ratios the solubility of the solid salt components, such as detergent builder salts, bleach, alkali metal silicates, and the like, is substantially increased since the presence of the potassium ( $K^+$ ) ions requires less water of hydration than the sodium ( $Na^+$ ) ions, such that more water is available to dissolve these salt compounds. Therefore, all or nearly all of the normally solid components are present dissolved in the aqueous phase. Since there is none or only a very low percentage, i.e. less than 5%, preferably less than 3% by weight, of suspended solids present in the formulation there is no or only reduced tendency for undissolved particles to settle out of the compositions causing, for example, formation of hard masses of particles, which could result in high bottle residues (i.e. loss of product). Furthermore, any undissolved solids tend to be present in extremely small particle sizes, usually colloidal or sub-colloidal, such as 1 micron or less, thereby further reducing the tendency for the undissolved particles to settle. Since there are substantially no undissolved solid particles of size in excess of 1 micron, the invention products tend to be transparent or at least translucent, depending, in part, on air bubble content.

A still further attribute of the invention compositions contributing to the overall product stability and low bottle residue is the high water absorption capacity of the cross-linked polyacrylic acid-type thickening agent. As a result of this high water absorption capacity virtually all of the aqueous vehicle component appears to be held tightly bound to the polymer matrix. Therefore, there appears to be no or substantially no free water present in the invention compositions. This apparent absence of free water (as well as the cohesiveness of the composition) is manifested by the observation that when the composition is poured from a bottle onto a piece of water absorbent filter paper virtually no water is absorbed onto the filter paper over a period of at least several hours or longer and, furthermore, the mass of the linear viscoelastic material poured onto the filter paper will retain its shape and structure. As a result of the absence of loosely bound water, there is virtually no phase separation between the aqueous phase and the polymeric matrix or dissolved solid particles. This characteristic is manifested by the fact that when the subject compositions are subjected to centrifugation, e.g. at 1000 rpm for 30 minutes, there is no phase separation and the composition remains homogeneous. The preferred compositions have remained stable for periods in excess of 6 months and more.

In our earlier application Serial No. 07/353,712, it was stated that to maximize physical (phase) stability, the density of the composition should be controlled such that the bulk density of the liquid phase is approximately the same as the bulk density of the entire composition, including the polymeric thickening agent. This control and equalization of the densities was achieved, according to our earlier invention, by providing the composition with a bulk density of at least 1.32 g/cc. A density of about 1.42 g/cc is essentially equivalent to zero air content.

It is important to note that the bulk density of the product can be adjusted by controlling the degree of aeration, as well as total solid content. Further, the dispersed air bubbles also contribute to the viscoelastic property of the product.

However, it has now been found that air bubble incorporation is not required to achieve stabilization and, in fact, we have been able to prepare stable thickened products with densities as low as 1.28 g/cc. At densities below about 1.28 cc/g, however, the flowability of the product tends to be degraded and the large air bubble content tends to cause the composition to be too highly translucent or cloudy to opaque. Therefore, the product density is preferably selected in the range of 1.28 to 1.42 g/cc, especially 1.32 to 1.40 g/cc, and most preferably from 1.35 to 1.40 g/cc. Within these ranges air bubble incorporation is determined to achieve the desired product aesthetic appearance and flow characteristics.

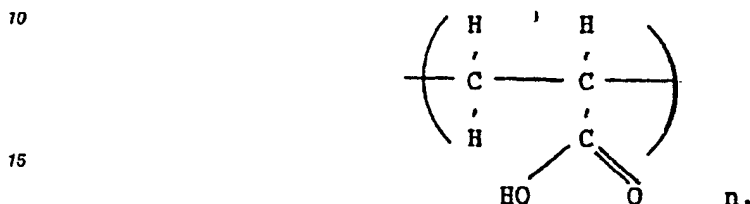
It has previously been found in connection with other types of thickened aqueous liquid, automatic dishwasher detergent compositions that agglomeration or escape of incorporated air bubbles could be avoided by incorporating certain surface active ingredients, especially higher fatty acids and the salts thereof, such as stearic acid, behenic acid, palmitic acid, sodium stearate, aluminum stearate, and the like.

Therefore, in the present invention, in order to avoid stabilization of air bubbles which may become incorporated into the compositions during normal processing, such as during various mixing steps, is avoided by post-adding the surface active ingredients to the remainder of the composition, under low shear conditions using mixing devices designed to minimize cavitation and vortex formation.

As will be described in greater detail below the surface active ingredients present in the composition will include the main detergent surface active cleaning agent, and will also preferably include anti-foaming

agent (e.g. phosphate ester) and higher fatty acid or salt thereof as a physical stabilizer.

Exemplary of the polycarboxylate type thickening agents are cross-linked polyacrylic acid-type thickening agents sold by B.F. Goodrich under their Carbopol trademark, including both the 900 series resins, especially Carbopol 941, which is the most ion-insensitive of this class of polymers, and Carbopol 940 and Carbopol 934, and the 600 series resins, especially Carbopol 614. The Carbopol 900 series resins are hydrophilic high molecular weight, cross-linked linear acrylic acid polymers having an average equivalent weight of 76, and the general structure illustrated by the following formula:



Carbopol 941 has a molecular weight of about 1,250,000; Carbopol 940 a molecular weight of approximately 4,000,000 and Carbopol 934 a molecular weight of approximately 3,000,000. The Carbopol 900 series resins are cross-linked with polyalkenyl polyether, e.g. about 1% of a polyallyl ether of sucrose having an average of about 5.8 allyl groups for each molecule of sucrose. The preparation of this class of cross-linked carboxylic polymers is described in U.S. Patent 2,798,053, the disclosure of which is incorporated by reference. Further detailed information on the Carbopol 900 series resins is available from B.F. Goodrich, see, for example, the B.F. Goodrich catalog GC-67, Carbopol® Water Soluble Resins.

In general, these thickening resins are preferably copolymers of a water dispersible copolymer of an alpha-beta monoethylenically unsaturated lower aliphatic carboxylic acid cross-linked with a polyether of a polyol selected from oligo saccharides, reduced derivatives thereof in which the carbonyl group is converted to an alcohol group and pentaerythritol, the hydroxyl groups of the polyol which are modified being etherified with allyl groups, there being preferably at least two such allyl groups per molecule.

These water-dispersible cross-linked thickening resins as described in the aforementioned U.S. Patent 2,798,053 and which have been commercialized by B.F. Goodrich as the Carbopol 900 series resins are prepared from essentially linear copolymers. More recently, B.F. Goodrich has introduced the Carbopol 600 series resin. These are high molecular weight, non-linear polyacrylic acid cross-linked with polyalkenyl ether. In addition to the non-linear or branched nature of these resins, they are also believed to be more highly cross-linked than the 900 series resins and have molecular weights between about 1,000,000 and 4,000,000.

Most especially useful of the Carbopol 600 series resins is Carbopol 614 which is the most chlorine bleach stable of this class of thickening resins. Carbopol 614 is also highly stable in the high alkalinity environment of the preferred liquid automatic dishwasher detergent compositions and is also highly stable to any anticipated storage temperature conditions from below freezing to elevated temperatures as high as 120° F, preferably 140° F, and especially 160° F, for periods of as long as several days to several weeks or months or longer.

While the most favorable results have now been achieved with Carbopol 614 polyacrylic resin, other linear or branched cross-linked polycarboxylate-type thickening agents can also be used in the compositions of this invention. As used herein "polycarboxylate-type" refers to water-soluble carboxyvinyl polymers of alpha,beta monoethylenically unsaturated lower aliphatic carboxylic acids, which may be linear or non-linear, and are exemplified by homopolymers of acrylic acid or methacrylic acid or water-dispersible or water-soluble salts, esters or amides thereof, or water-soluble copolymers of these acids or their salts, esters or amides with each other or with one or more other ethylenically unsaturated monomers, such as, for example, styrene, maleic acid, maleic anhydride, 2-hydroxyethylacrylate, acrylonitrile, vinyl acetate, ethylene, propylene, and the like, and which have molecular weights of from about 500,000 to 10,000,000 and are cross-linked or interpolymers with a multi-vinyl or multi-allylic functionalized cross-linking agent, especially with a polyalkenyl ether of a polyhydric compound.

These homopolymers or copolymers are characterized by their high molecular weight, in the range of from about 500,000 to 10,000,000, preferably 750,000 to 5,000,000, especially from about 1,000,000 to 4,000,000, and by their water solubility, generally at least to an extent of up to about 5% by weight, or more, in water at 25° C.



These thickening agents are used in their cross-linked form wherein the cross-linking may be accomplished by means known in the polymer arts, as by irradiation, or, preferably, by the incorporation into the monomer mixture to be polymerized of known chemical cross-linking monomeric agents, typically polyunsaturated (e.g. diethylenically unsaturated) monomers, such as, for example, divinylbenzene, divinylether of diethylene glycol, N,N'-methylene-bisacrylamide, polyalkenylpolyethers (such as described above), and the like. Typically, amounts of cross-linking agent to be incorporated in the final polymer may range from about 0.01 to about 5 percent, preferably from about 0.05 to about 2 percent, and especially, preferably from about 0.1 to about 1.5 percent, by weight of cross-linking agent to weight of total polymer. Generally, those skilled in the art will recognize that the degree of cross-linking should be sufficient to impart some coiling of the otherwise generally linear or non-linear polymeric compound while maintaining the cross-linked polymer at least water dispersible and highly water-swellaable in an ionic aqueous medium.

The amount of the high molecular weight, cross-linked polyacrylic acid or other high molecular weight, hydrophilic cross-linked polycarboxylate thickening agent to impart the desired rheological property of linear viscoelasticity will generally be in the range of from about 0.1 to 2%, preferably from about 0.2 to 1.4%, by weight, based on the weight of the composition, although the amount will depend on the particular cross-linking agent, ionic strength of the composition, hydroxyl donors and the like.

The compositions of this invention must include sufficient amount of potassium ions and sodium ions to provide a weight ratio of K/Na of at least 1:1, preferably from 1:1 to 45:1, especially from about 1:1 to 4:1, more preferably from 1.05:1 to 3:1, such as 1.1:1, 1.2:1, 1.5:1, or 2:1. When the K/Na ratio is less than 1 there is insufficient solubility of the normally solid ingredients whereas when the K/Na ratio is more than 45, especially when it is greater than about 3 or 4, the product becomes too liquid and phase separation begins to occur. When the K/Na ratios become much larger than 45, such as in an all or mostly potassium formulation, the polycarboxylate thickener loses its absorption capacity and begins to salt out of the aqueous phase.

The potassium and sodium ions will be made present in the compositions as the alkali metal cation of the detergent builder salt(s), as well as alkali metal silicate or alkali metal hydroxide components of the compositions. The alkali metal cation may also be present in the compositions as a component of anionic detergent, bleach or other ionizable salt compound additive, e.g. alkali metal carbonate. In determining the K/Na weight ratios all of these sources should be taken into consideration.

Specific examples of detergent builder salts include the polyphosphates, such as alkali metal pyrophosphate, alkali metal tripolyphosphate, alkali metal metaphosphate, and the like, for example, sodium or potassium tripolyphosphate (hydrated or anhydrous), tetrasodium or tetrapotassium pyrophosphate, sodium or potassium hexa-metaphosphate, trisodium or tripotassium orthophosphate and the like, sodium or potassium carbonate, sodium or potassium citrate, sodium or potassium nitrilotriacetate, and the like.

In accordance with the present invention, however, the detergent builder salts will be comprised of mixtures of at least potassium tripolyphosphate (KTPP) and sodium tripolyphosphate (NaTPP) (especially hexahydrate). Typical ratios of KTPP to NaTPP are from about 1.4:1 to 10:1, especially from about 2:1 to 8:1. The total amount of detergent builder salts is preferably from about 10 to 35% by weight, more preferably from about 15 to 35%, especially from about 18 to 30% by weight of the composition. Of this total amount of the detergent builders at least 50% by weight (preferably at least about 8% by weight of the composition) will be KTPP and preferably at least 5% by weight (preferably at least 2% by weight of the composition) will be NaTPP. More preferably, the alkali metal detergent builder salt will be comprised of from about 65 to 95% of KTPP, especially 75 to 90% of KTPP and from about 5 to 35%, especially 10 to 25% of NaTPP. In terms of the total composition, the amount of KTPP will be in the range of from about 8 to 25%, preferably 15 to 22% by weight, and the amount of NaTPP will be in the range of from about 2 to 10%, preferably 3 to 8% by weight.

When other alkali metal detergent builder salts are present in the formulation, they will usually be present in amounts less than 5% by weight based on the total composition and, in any case, in amounts to maintain the K/Na ratios to within the above described range, preferably from 1:1.1 to 1:3.

The linear viscoelastic compositions of this invention may, and preferably will, contain a small, but stabilizing effective amount of a long chain fatty acid or monovalent or polyvalent salt thereof. Although the manner by which the fatty acid or salt contributes to the rheology and stability of the composition has not been fully elucidated it is hypothesized that it may function as a hydrogen bonding agent or cross-linking agent for the polymeric thickener.

The preferred long chain fatty acids are the higher aliphatic fatty acids having from about 10 to 50 carbon atoms, more preferably from about 12 to 40 carbon atoms, and especially preferably from about 14 to 40 carbon atoms, inclusive of the carbon atom of the carboxyl group of the fatty acid. The aliphatic radical may be saturated or unsaturated and may be straight or branched. Straight chain saturated fatty

acids are preferred. Mixtures of fatty acids may be used, such as those derived from natural sources, such as tallow fatty acid, coco fatty acid, soya fatty acid, etc., or from synthetic sources available from industrial manufacturing processes.

Thus, examples of the fatty acids include, for example, decanoic acid, dodecanoic acid, palmitic acid, myristic acid, stearic acid, behenic acid, oleic acid, eicosanoic acid, tallow fatty acid, coco fatty acid, soya fatty acid, mixtures of these acids, etc. Stearic acid and mixed fatty acids, e.g. stearic acid/palmitic acid, are preferred.

It has, however, also recently been discovered by some of us and others that further improvements in phase stability, particularly under elevated temperature storage conditions, and maintenance of product viscosity levels can be obtained by using longer chain length fatty acids in the range of from C<sub>18</sub> to C<sub>40</sub>. Either individual or mixtures of these longer chain length fatty acids can be used, however, the average chain length should be in the range of from about 20 to 32 carbon atoms, especially 24 to 30 carbon atoms and mixture of fatty acids encompassing this range are preferred. Suitable mixed fatty acids are commercially available, for instance those sold under the trade name Syncrowax by Croda.

When the free acid form of the fatty acid is used directly it will generally associate with the potassium and sodium ions in the aqueous phase to form the corresponding alkali metal fatty acid soap. However, the fatty acid salts may be directly added to the composition as sodium salt or potassium salt, or as a polyvalent metal salt, although the alkali metal salts of the fatty acids are preferred fatty acid salts.

The preferred polyvalent metals are the di- and trivalent metals of Groups IIA, IIB and IIIB, such as magnesium, calcium, aluminum and zinc, although other polyvalent metals, including those of Groups IIIA, IVA, VA, VB, VIB, VIIB and VIII of the Periodic Table of the Elements can also be used. Specific examples of such other polyvalent metals include Ti, Zr, V, Nb, Mn, Fe, Co, Ni, Cd, Sn, Sb, Bi, etc. Generally, the metals may be present in the divalent to pentavalent state. Preferably, the metal salts are used in their higher oxidation states. Naturally, for use in automatic dishwashers, as well as any other applications where the invention composition will or may come into contact with articles used for the handling, storage or serving of food products or which otherwise may come into contact with or be consumed by people or animals, the metal salt should be selected by taking into consideration the toxicity of the metal. For this purpose, the alkali metal and calcium and magnesium salts are especially higher preferred as generally safe food additives.

The amount of the fatty acid or fatty acid salt stabilizer to achieve the desired enhancement of physical stability will depend on such factors as the nature of the fatty acid or its salt, the nature and amount of the thickening agent, detergent active compound, inorganic salts, other ingredients, as well as the anticipated storage and shipping conditions.

Generally, however, amounts of the fatty acid or fatty acid salt stabilizing agents in the range of from about 0.02 to 2%, preferably 0.04 to 1%, more preferably from about 0.06 to 0.8%, especially preferably from about 0.08 to 0.4%, provide a long term stability and absence of phase separation upon standing or during transport at both low and elevated temperatures as are required for a commercially acceptable product.

Depending on the amounts, proportions and types of fatty acid physical stabilizers and polycarboxylate thickening agents, the addition of the fatty acid or salt not only increases physical stability but also provides a simultaneous increase in apparent viscosity. Amounts of fatty acid or salt to polymeric thickening agent in the range of from about 0.08-0.4 weight percent fatty acid salt and from about 0.4-1.5 weight percent polymeric thickening agent are usually sufficient to provide these simultaneous benefits and, therefore, the use of these ingredients in these amounts is most preferred.

In order to achieve the desired benefit from the fatty acid or fatty acid salt stabilizer, without stabilization of excess incorporated air bubbles and consequent excessive lowering of the product bulk density, the fatty acid or salt is preferably post-added to the formulation, preferably together with the other surface active ingredients, including detergent active compound and anti-foaming agent, when present. These surface active ingredients are preferably added as an emulsion in water wherein the emulsified oily or fatty materials are finely and homogeneously dispersed throughout the aqueous phase. To achieve the desired fine emulsification of the fatty acid or fatty acid salt and other surface active ingredients, it is usually necessary to heat the emulsion (or preheat the water) to an elevated temperature near the melting temperature of the fatty acid or its salt. For example, for stearic acid having a melting point of 68°-69° C, a temperature in the range of between 50° C and 70° C will be used. For lauric acid (m.p. = 47° C) an elevated temperature of about 35° to 50° C can be used. Apparently, at these elevated temperatures the fatty acid or salt and other surface active ingredients can be more readily and uniformly dispersed (emulsified) in the form of fine droplets throughout the composition.

Foam inhibition is important to increase dishwasher machine efficiency and minimize destabilizing

effects which might occur due to the presence of excess foam within the washer during use. Foam may be reduced by suitable selection of the type and/or amount of detergent active material, the main foam-producing component. The degree of foam is also somewhat dependent on the hardness of the wash water in the machine whereby suitable adjustment of the proportions of the builder salts, such as NaTPP which  
 5 has a water softening effect, may aid in providing a degree of foam inhibition. However, it is generally preferred to include a chlorine bleach stable foam depressant or inhibitor. Particularly effective are the alkyl phosphoric acid esters of the formula



15 and especially the alkyl acid phosphate esters of the formula



In the above formulas, one or both R groups in each type of ester may represent independently a C<sub>12</sub>-C<sub>20</sub>  
 25 alkyl group. The ethoxylated derivatives of each type of ester, for example, the condensation products of one mole of ester with from 1 to 10 moles, preferably 2 to 6 moles, more preferably 3 or 4 moles, ethylene oxide can also be used. Some examples of the foregoing are commercially available, such as the products SAP from Hooker and LPKN-158 from Knapsack. Mixtures of the two types, or any other chlorine bleach stable types, or mixtures of mono- and diesters of the same type, may be employed. Especially preferred is  
 30 a mixture of mono- and di-C<sub>15</sub>-C<sub>18</sub> alkyl acid phosphate esters such as monostearyl/distearyl acid phosphates 1.2/1, and the 3 to 4 mole ethylene oxide condensates thereof. When employed, proportions of 0.05 to 1.5 weight percent, preferably 0.1 to 0.5 weight percent, of foam depressant in the composition is typical, the weight ratio of detergent active component (d) to foam depressant (e) generally ranging from about 10:1 to 1:1 and preferably about 5:1 to 1:1. Other defoamers which may be used include, for  
 35 example, the known silicones, such as available from Dow Chemicals. In addition, it is an advantageous feature of this invention that many of the stabilizing salts, such as the stearate salts, for example, aluminum stearate, when included, are also effective as foam killers.

Although any chlorine bleach compound may be employed in the compositions of this invention, such as dichloro-isocyanurate, dichloro-dimethyl hydantoin, or chlorinated TSP, alkali metal or alkaline earth  
 40 metal, e.g. potassium, lithium, magnesium and especially sodium, hypochlorite is preferred. The composition should contain sufficient amount of chlorine bleach compound to provide about 0.2 to 4.0% by weight of available chlorine, as determined, for example, by acidification of 100 parts of the composition with excess hydrochloric acid. A solution containing about 0.2 to 4.0% by weight of sodium hypochlorite contains or provides roughly the same percentage of available chlorine. About 0.8 to 1.6% by weight of available  
 45 chlorine is especially preferred. For example, sodium hypochlorite (NaOCl) solution of from about 11 to about 13% available chlorine in amounts of about 3 to 20%, preferably about 7 to 12%, can be advantageously used.

Another surprising and unexpected benefit and advantage of the preferred Carbopol thickened compositions of this invention is the essentially total masking of any chlorine bleach odor which is characteristic of,  
 50 for example, the prior known clay thickened products. By virtue of the masking of chlorine bleach odor, it has been found that the subject compositions can be blended with substantially lower amounts of fragrance, e.g. lemon oil fragrance, to achieve the same or superior olfactory sensation for the consumer. For instance, as little as 0.05% of lemon fragrance will have the same effect as about 0.12% in clay thickened products.

Furthermore, the addition of fragrance does not adversely affect the stability criteria, as previously  
 55 defined, such as viscosity or phase stability, of the compositions.

Detergent active material useful herein should be low-foaming and stable in the presence of chlorine bleach, when present, especially hypochlorite bleach, and for this purpose those of the organic anionic, nonionic, amine oxide, phosphine oxide, sulphoxide or betaine water dispersible surfactant types are

preferred, the first mentioned anionics being most preferred. Particularly preferred surfactants herein are the linear or branched alkali metal mono- and/or di-(C<sub>8</sub>-C<sub>14</sub>) alkyl diphenyl oxide mono- and/or di-sulphates, commercially available for example as DOWFAX (registered trademark) 3B-2 and DOWFAX 2A-1. In addition, the surfactant should be compatible with the other ingredients of the composition. Other suitable organic anionic, non-soap surfactants include the primary alkylsulphates, alkylsulphonates, alkylarylsulphonates and sec.-alkylsulphates. Examples include sodium C<sub>10</sub>-C<sub>18</sub> alkylsulphates such as sodium dodecylsulphate and sodium tallow alcoholsulphate; sodium C<sub>10</sub>-C<sub>18</sub> alkanesulphonates such as sodium hexadecyl-1-sulphonate and sodium C<sub>12</sub>-C<sub>18</sub> alkylbenzenesulphonates such as sodium dodecylbenzenesulphonates. The corresponding potassium salts may also be employed.

As other suitable surfactants or detergents, the amine oxide surfactants are typically of the structure R<sub>2</sub>R'NO, in which each R represents a lower alkyl group, for instance, methyl, and R' represents a long chain alkyl group having from 8 to 22 carbon atoms, for instance a lauryl, myristyl, palmityl or cetyl group. Instead of an amine oxide, a corresponding surfactant phosphine oxide R<sub>2</sub>R'PO or sulphoxide RR'SO can be employed. Betaine surfactants are typically of the structure R<sub>2</sub>R'N<sup>+</sup>R''COO<sup>-</sup>, in which each R represents a lower alkylene group having from 1 to 5 carbon atoms. Specific examples of these surfactants include lauryl-dimethylamine oxide, myristyl-dimethylamine oxide, the corresponding phosphine oxides and sulphoxides, and the corresponding betaines, including dodecyldimethylammonium acetate, tetradecyldiethylammonium pentanoate, hexadecyldimethylammonium hexanoate and the like. For biodegradability, the alkyl groups in these surfactants should be linear, and such compounds are preferred.

Surfactants of the foregoing type, all well known in the art, are described, for example, in U.S. Patents 3,985,668 and 4,271,030. If chlorine bleach is not used than any of the well known low-foaming nonionic surfactants such as alkoxylated fatty alcohols, e.g. mixed ethylene oxide-propylene oxide condensates of C<sub>8</sub>-C<sub>22</sub> fatty alcohols can also be used.

The chlorine bleach stable, water dispersible organic detergent-active material (surfactant) will normally be present in the composition in minor amounts, generally about 1% by weight of the composition, although smaller or larger amounts, such as up to about 5%, such as from 0.1 to 5%, preferably from 0.3 or 0.4 to 2% by weight of the composition, may be used.

Alkali metal (e.g. potassium or sodium) silicate, which provides alkalinity and protection of hard surfaces, such as fine china glaze and pattern, is generally employed in an amount ranging from about 5 to 20 weight percent, preferably about 5 to 15 weight percent, more preferably 8 to 12% in the composition. The sodium or potassium silicate is generally added in the form of an aqueous solution, preferably having Na<sub>2</sub>O:SiO<sub>2</sub> or K<sub>2</sub>O:SiO<sub>2</sub> ratio of about 1:1.3 to 1:2.8, especially preferably 1:2.0 to 1:2.6. At this point, it should be mentioned that many of the other components of this composition, especially alkali metal hydroxide and bleach, are also often added in the form of a preliminary prepared aqueous dispersion or solution. However, unless otherwise noted, when amounts of a particular ingredient are given, the reference is to an active ingredient basis, i.e. does not include the aqueous carrier.

In addition to the detergent active surfactant, foam inhibitor, alkali metal silicate corrosion inhibitor, and detergent builder salts, which all contribute to the cleaning performance, it is also known that the effectiveness of the liquid automatic dishwasher detergent compositions is related to the alkalinity, and particularly to moderate to high alkalinity levels. Accordingly, the compositions of this invention will have pH values of at least about 9.5, preferably at least about 11 to as high as 14, generally up to about 13 or more, and, when added to the aqueous wash bath at a typical concentration level of about 10 grams per liter, will provide a pH in the wash bath of at least about 9, preferably at least about 10, such as 10.5, 11, 11.5 or 12 or more.

The alkalinity will be achieved, in part, by the alkali metal ions contributed by the alkali metal detergent builder salts, e.g. sodium tripolyphosphate, potassium tripolyphosphate, and alkali metal silicate, however, it is usually necessary to include alkali metal hydroxide, e.g. NaOH or KOH, to achieve the desired high alkalinity. Amounts of alkali metal hydroxide in the range of from about 0.5 to 8%, preferably from 1 to 6%, more preferably from about 1.2 to 4%, by weight of the composition will be sufficient to achieve the desired pH level and/or to adjust the K/Na weight ratio.

While as stated above, high alkalinity is desired with regard to improved cleaning performance, it has now additionally been discovered that the pH of the composition during processing is highly important for product stability, e.g. viscosity, and bleach stability and phase stability. Specifically, it has been found that, especially for the Carbopol class of thickening agents, the amount of basic neutralizing agent, preferably NaOH, should be sufficient to provide the neutralized thickening agent with a pH of at least 11, preferably at least 11.5. Furthermore, these high alkalinity levels should be maintained throughout the product formulation steps, namely, the pH value of at least 11, preferably at least 11.5, should be maintained during the successive additions of the other ingredients of the composition.

As will be shown by the examples given below, if the pH drops to a lower value, there is a loss in viscosity, as well as stability (as manifested by liquid phase separation) over time.

Other alkali metal salts, such as alkali metal carbonate may also be present in the compositions in minor amounts, for example from 0 to 4%, preferably 0 to 2%, by weight of the composition.

5 Another often beneficial additive for the present liquid automatic dishwasher detergent compositions is a relatively low molecular weight, non-crosslinked polyacrylic acid or neutralized with caustic, such as the commercial product Acrysol LMW=45N, which has a molecular weight of about 4,500. The low m.w. polyacrylic acids can provide additional thickening characteristics but are primarily introduced for their ability to function as a builder or chelating agent. In this capacity, the low m.w. polyacrylic acids can  
10 contribute to reduced spotting or streaking and reduced filming on dishes, glassware, pots, pans, and other utensils and appliances.

Generally, a suitable molecular weight ranges for the non-crosslinked polyacrylic acid is from about 800 to 200,000, preferably 1000 to 150,000, especially from about 2,000 to 100,000. When present in the formulation, the non-crosslinked polyacrylic acid can be used in amounts up to about 10% by weight,  
15 preferably from about 1 to 8% by weight, especially 2 to 6% by weight of the composition.

Other conventional ingredients may be included in these compositions in small amounts, generally less than about 3 weight percent, such as perfume, hydrotropic agents such as the sodium benzene, toluene, xylene and cumene sulphonates, preservatives, dyestuffs and pigments and the like, all of course being stable to chlorine bleach compound and high alkalinity. Especially preferred for coloring are the chlorinated  
20 phthalocyanines and polysulphides of aluminosilicate which provide, respectively, pleasing green and blue tints. To achieve stable yellow colored products, the bleach stable mixed dyes C.I. Direct Yellow 28 (C.I. 19555) or C.I. Direct Yellow 29 (C.I. 19556) can be added to the compositions. These colors meet stability criteria described in Table A.  $\text{TiO}_2$  may be employed for whitening or neutralizing off-shades.

Although for the reasons previously discussed excessive air bubbles are not often desirable in the invention compositions, depending on the amounts of dissolved solids and liquid phase densities, incorporation of small amounts of finely divided air bubbles, generally up to about 10% by volume, preferably up to about 4% by volume, more preferably up to about 2% by volume, can be incorporated to adjust the visual appearance, product density and flowability. The incorporated air bubbles should be finely divided, such as up to about 100 microns in diameter, preferably from about 20 to about 40 microns in diameter.

30 Other inert gases can also be used, such as nitrogen, carbon dioxide, helium, oxygen, etc.

The amount of water contained in these compositions should, of course, be neither so high as to produce unduly low viscosity and fluidity, nor so low as to produce unduly high viscosity and low flowability, linear viscoelastic properties in either case being diminished or destroyed by increasing tan  $\delta$ . Such amount is readily determined by routine experimentation in any particular instance, and generally will range from 30  
35 to 75 weight percent, preferably about 35 to 65 weight percent. Preferably, the water should also be deionized or softened.

The manner of formulating the invention compositions is also important. As discussed above, the order of mixing the ingredients as well as the manner in which the mixing is performed will generally have a significant effect on the properties of the composition, and in particular on product density, (by incorporation  
40 of more or less air), viscosity and physical stability (e.g. phase separation). Thus, according to the preferred practice of this invention the compositions are prepared by forming a dispersion of the polycarboxylate type thickener in heated water, e.g. 35° to 60° C (95° to 140° F), preferably 40° to 50° C (104° to 122° F), under moderate to high shear conditions, neutralizing the dissolved polymer to a pH of at least 11, preferably at least 11.5, such as from 11.5 to 13.0, to cause gelation. After transferring the thickener  
45 dispersion to a main mixing tank processing is continued by introducing, while continuing mixing, the detergent builder salts, alkali metal silicates, chlorine bleach compound and remaining detergent additives, including any previously unused alkali metal hydroxide, if any, other than the surface-active compounds. All of the additional ingredients can be added simultaneously or sequentially. Preferably, the ingredients are added sequentially, with mixing continued for from 2 to 10 minutes for each ingredient, although it is not  
50 necessary to complete the addition of one ingredient before beginning to add the next ingredient. Furthermore, one or more of these ingredients can be divided into portions and added at different times. These mixing steps should also be performed under moderate to high shear rates to achieve complete and uniform mixing. These additional ingredient mixing steps may be carried out at room temperature, but preferably the elevated temperature of the thickener slurry is maintained. The composition may be allowed  
55 to age, if necessary, to cause dissolved or dispersed air to dissipate out of the composition.

The remaining surface active ingredients, including the anti-foaming agent, organic detergent compound, and fatty acid or fatty acid salt stabilizer is post-added to the previously formed mixture in the form of an aqueous emulsion (using from about 1 to 10%, preferably from about 2 to 4% of the total water added

to the composition other than water added as carrier for other ingredients or water of hydration) which is pre-heated to a temperature in the range of from about  $T_m - 5$  to  $T_m + 20$ , preferably from about  $T_m$  to  $T_m + 10$ , where  $T_m$  is the melting point temperature of the fatty acid or fatty acid salt. For the stearic acid stabilizer the heating temperature is in the range of  $150^\circ$  to  $170^\circ$  F ( $65^\circ$  to  $77^\circ$  C). For the high chain length fatty acids and mixtures  $C_{18}$ - $C_{36}$ , correspondingly higher temperatures may be used, such as from about  $160^\circ$  to  $200^\circ$  F (ca.  $70^\circ$  to  $95^\circ$  C). However, if care is taken to avoid excessive air bubble incorporation during the gelation step or during the mixing of the detergent builder salts and other additives, for example, by operating under vacuum, or using low shearing conditions, or special mixing apparatus, etc., the order of addition of the surface active ingredients should be less important.

In accordance with an especially preferred embodiment, the thickened linear viscoelastic aqueous automatic dishwasher detergent composition of this invention includes, on a weight basis:

(a)

(i) 8 to 25%, preferably 10 to 20%, potassium triphosphate detergent builder;

(ii) 2 to 10%, preferably 4 to 8%, sodium triphosphate detergent builder, at an (i)/(ii) weight ratio

of from about 1.4/1 to 10/1, preferably 2/1 to 6/1;

(b) 5 to 15, preferably 8 to 12%, alkali metal silicate;

(c) 1 to 6%, preferably 1.2 to 4%, alkali metal hydroxide;

(d) 0.1 to 3%, preferably 0.5 to 2%, chlorine bleach stable, water-dispersible, low-foaming organic detergent active material, preferably non-soap anionic detergent;

(e) 0.05 to 1.5%, preferably 0.1 to 0.5%, chlorine bleach stable foam depressant;

(f) chlorine bleach compound in an amount to provide about 0.2 to 4%, preferably 0.8 to 1.6%, of available chlorine;

(g) non-linear, water-dispersible polyacrylic acid thickening agent comprising at least one high molecular weight hydrophilic polycarboxylate having a molecular weight of from 750,000 to 4,000,000, preferably 800,000 to 3,000,000, in an amount to provide a linear viscoelasticity to the formulation, preferably from about 0.2 to 2%, especially preferably from about 0.4 to 1.5%, more preferably from about 0.4 to 1.0%;

(h) a long chain fatty acid or a metal salt of a long chain fatty acid in an amount effective to increase the physical stability of the compositions, preferably from 0.08 to 0.4%, more preferably from 0.1 to 0.3%; and

(i) 0 to 10%, preferably 1 to 8%, especially 2 to 6% of non-crosslinked polyacrylic acid having a molecular weight in the range of from about 800 to 200,000, preferably 1000 to 150,000, especially 2,000 to 100,000; and

(j) balance water, preferably from about 30 to 75%, more preferably from about 35 to 65%; and wherein in the entire composition the ratio, by weight, of potassium ions to sodium ions is from about 1.05/1 to 3/1 or 4/1, preferably from 1.1/1 to 2.5/1. The compositions may also have an amount of air incorporated therein such that the bulk density of the composition is from about 1.28 to 1.42 g/cc, preferably from about 1.32 to 1.42 g/cc, more preferably from about 1.35 to 1.40 g/cc.

The compositions will be supplied to the consumer in suitable dispenser containers preferably formed of molded plastic, especially polyolefin plastic, and most preferably opaque or translucent polyethylene, for which the invention compositions appear to have particularly favorable slip characteristics. In addition to their linear viscoelastic character, the compositions of this invention may also be characterized as pseudoplastic gels (non-thixotropic) which are typically near the borderline between liquid and solid viscoelastic gel, depending, for example, on the amount of the polymeric thickener. The invention compositions can be readily poured from their containers without any shaking or squeezing, i.e. have a sufficiently low yield stress value to flow under their own weight (gravity), although squeezable containers are often convenient and accepted by the consumer for gel-like products.

The liquid aqueous linear viscoelastic automatic dishwasher compositions of this invention are readily employed in known manner for washing dishes, other kitchen utensils and the like in an automatic dishwasher, provided with a suitable detergent dispenser, in an aqueous wash bath containing an effective amount of the composition, generally sufficient to fill or partially fill the automatic dispenser cup of the particular machine being used.

The invention also provides a method for cleaning dishware in an automatic dishwashing machine with an aqueous wash bath containing an effective amount of the liquid linear viscoelastic automatic dishwasher detergent composition as described above. The composition can be readily poured from the polyethylene container with little or no squeezing or shaking into the dispensing cup of the automatic dishwashing machine and will be sufficiently viscous and cohesive to remain securely within the dispensing cup until shear forces are again applied thereto, such as by the water spray from the dishwashing machine.

The invention may be put into practice in various ways and a number of specific embodiments will be

described to illustrate the invention with reference to the accompanying examples.

All amounts and proportions referred to herein are by weight of the composition unless otherwise indicated.

5 Example 1

The following formulations A-F were prepared as described below:

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Table 1

FORMULATION						
INGREDIENT	A	B	C	D	E	F
WATER	Q.A	Q.A	Q.A	Q.A	Q.A	Q.A
CARBOPOL 941	0.9	-	-	-	-	-
CARBOPOL 940	-	0.9	-	-	-	-
CARBOPOL 614	-	-	0.9	0.9	0.9	0.9
NaOH (50%)	2.4	4.5	4.5	4.0	4.5	4.5
Na-SILICATE (47.5%)(1:2.4)	21	21	20.83	20.83	20.83	20.83
TKPP	15	15	-	-	-	-
KTPP	-	-	20.35	20.35	13	20.35
NaTPP (ANHYDROUS)	13	13	5.26	5.26	3	5.26
DOWFAX 3B2	1	0.8	0.8	0.8	0.8	0.8
LPKN (ANTI-FOAMING AGENT)	0.16	0.16	0.16	0.16	0.16	0.16
FATTY ACID	0.10(2)	0.20(1)	0.15(3)	0.15(2)	0.15(2)	0.15(2)
BLEACH (13.1%)	8.1	11.1	10.13	10.13	10.13	10.13
GRAPHOL GREEN	0.0025	0.003	0.003	0.003	0.003	-
CI DIRECT YELLOW 28	-	-	-	-	-	0.003
AIR (Vol.%) APPROX.	2	2	2	2	2	2
ACRYSOL LMW 45-N (45.0%)	-	-	-	-	4.4	-



Table 1 (cont.)

FORMULATION						
INGREDIENT	A	B	C	D	E	F
HIGHLIGHTS (FRAGRANCE)	-	-	0.05	0.05	0.05	0.05
K/Na	0.98	0.98	1.61	- 1.61	1.17	1.61
DENSITY	1.35	1.37	1.37	1.37	1.28	1.37
STABILITY AMBIENT	8 WKS	8 WKS	24 WKS	24 WKS	12 WKS	4 WKS
STABILITY 100°F	2 WKS	2 WKS	-	20 WKS	8 WKS	4 WKS
STABILITY 120°F	-	-	-	8 WKS	8 WKS	4 WKS
STABILITY 140°F	-	-	-	2 WKS	2 WKS	2 WKS
CRYSTAL GROWTH (100°F)	YES	YES	NO	NO	NO	NO
RHEOGRAM	FIG.1	FIG.2	FIG.3	FIG.4	FIG.5	FIG.6

- (1) Syncrowax C24-26  
 (2) Stearic Acid  
 (3) Syncrowax C18-36

Formulations A, B, C, D, E and F are prepared by first forming a uniform dispersion of the Carbopol 614 or 940 thickener in about 97% of the water of the total formula water. The Carbopol is slowly added by sprinkling it into the vortex of previously colored deionized water preheated to a temperature of 105°F using a mixer equipped with a premier blade, with agitation set at a medium shear rate, as recommended by the manufacturer. After mixing for about 15 minutes, the dispersion is then neutralized by addition, under the same mixing, of the caustic soda (50% NaOH) component until a thickened product of gel-like

consistency is formed (about 10 minutes).

To the resulting gelled dispersion the silicate, sodium tripolyphosphate (NaTPP), tetrapotassium pyrophosphate (TKPP), or potassium tripolyphosphate (KTPP), the surfactant emulsion (described below) and bleach and color, are added sequentially, in the order stated, with the mixing continued at medium shear for several minutes before adding the next ingredient. After the addition of the surfactant emulsion (at 160° F), the mixture is cooled to from 90° -95° F before the bleach is added.

Separately, the surfactant emulsion of the phosphate anti-foaming agent (LPKN), stearic acid or fatty acid mixture and detergent (Dowfax 3B2) is prepared by adding these ingredients to the remaining 3% of water and heating the resulting mixture to a temperature in the range of 160° F (71° C). In formulation E, the Acrysol LMW 45-N may be added at this stage.

The rheograms for the formulations A, B, C, D, E and F are shown in figures 1-6, respectively.

These rheograms are obtained with the System 4 Rheometer from Rheometrics equipped with a Fluid Servo with a 100 grams-centimeter torque transducer and a 50 millimeter parallel plate geometry having an 0.8 millimeter gap between plates. All measurements are made at room temperature (25° ±1° C) in a humidity chamber after a 5 minute or 10 minute holding period of the sample in the gap. The measurements are made by applying a frequency of 10 radians per second.

All of the composition formulations C, D and F exhibit linear viscoelasticity as seen from the rheograms of figure 2-6. No phase separation at from ambient temperature to 140° F were observed for any of the formulations for at least the minimum number of weeks required to satisfy the criteria stability as shown in Table A above. Formulations E and F were still being tested when this application was filed.

However, in the control formulations A and B maintained at 100° F, the TKPP crystallized in the aqueous phase and eventually formed sufficiently large size crystals which separated to the bottom of the composition. Also, as seen in figures 1 and 2 formulations A and B are not linear viscoelastic, at least within the preferred criteria as previously described. Formulations C, D, E and F, according to the invention did not undergo any crystal growth.

For the bottle residue test, each formulation is allowed to age for about 1 week at ambient temperature in a standard 32 ounce small necked polyethylene bottle. An amount of product is poured from the bottle to fill a standard sized dispenser cup of an automatic dishwasher. The bottle is then replaced in an upright position and is retained in the upright position for at least 15 minutes. This procedure of filling the dispenser cup, placing the container in the upright position and waiting at least 15 minutes is repeated until no more product flows from the bottle. At this time, the weight of the bottle is measured. Bottle residue is calculated as

$$\frac{W_f}{W_o} \times 100$$

Wo is the initial weight of the filled bottle and Wf is the final weight of the filled bottle. The bottle residue for each formulation A-F is about 4 to 5%. Formulations C-F have viscosities of from 10,000 to 20,000 measured with a Brookfield LVT viscometer, #4 spindle at 20 rpm measured at 80° F. all of these products are easily pourable from the polyethylene bottle.

#### Example 2

A Carbopol 614 slurry is formed as described in Example 1 except that the coloring agent is first added to the deionized water (about 92% of the total added water) and the amounts of the ingredients are changed as shown below. The premix (surfactant emulsion) of the surface active ingredients is also formed as in Example 1 using stearic acid as the fatty acid stabilizer and the remaining 8% of the total added water.

The ingredients are then mixed together with the Carbopol 614 slurry in the following order: alkali metal silicate, NaTPP (powder), KTPP (powder), surfactant emulsion, bleach and perfume. The resulting composition is obtained with the following ingredients in the following amounts:

	<u>Ingredient</u>	<u>Amount (wt%)</u>
	Deionized Water	q.s. 100
5	Carbopol 614	1.00
	NaOH (38%Na <sub>2</sub> O)	6.38
	Na silicate (1:24)(47.5%)	20.83
	KTPP (anhydrous) powder	20.35
	NaTPP (3% H <sub>2</sub> O) powder	5.26
10	Dowfax 3B2	0.80
	LPKN	0.16
	Stearic Acid	0.15
	Bleach (Na hypochlorite-13%)	9.23
	CI Pigment Green 7 (CI 74260)	0.0024
15	Highlights (fragrance)	0.05

The composition has a pH of 11.3 + 0.2 and density (sp.gr.) of 1.39 + 0.03. The viscosity at 80° F measured with a Brookfield LVT viscometer at 20 rpm with a #4 spindle is 12,000 + 2,000.

All of the preferred criteria as set forth in Table A above are satisfied.

20

#### Claims

1. A linear viscoelastic aqueous liquid automatic dishwasher detergent composition comprising water, up to about 2% by weight of long chain fatty acid or salt thereof, from about 0.1 to 5% by weight of low-foaming chlorine bleach stable, water dispersible automatic dishwasher non-soap organic detergent, from about 5 to 40% by weight of alkali metal detergent builder salt, at least about 50% by weight of which is potassium triphosphate, up to about 20% by weight of a chlorine bleach compound, and cross-linked polycarboxylate-type thickening agent having a molecular weight of at least about 800,000 wherein the aqueous phase includes both sodium and potassium ions at a K/Na weight ratio of from about 1/1 to about 45/1.
2. The composition of claim 1 wherein the long chain fatty acid or salt thereof is present in an amount of from about 0.01 to 0.8% by weight.
3. The composition of claim 1 which further comprises an alkali metal silicate anti-corrosion agent.
4. The composition of claim 1 which further comprises up to about 2% by volume, based on the total volume of the composition, of air in the form of finely dispersed bubbles.
5. The composition of claim 1 wherein said polycarboxylate-type thickening agent is a cross-linked polyacrylic acid having a molecular weight in the range of from about 1,000,000 to about 4,000,000.
6. The composition of claim 5 wherein the cross-linked polyacrylic acid thickening agent is present in an amount of from about 0.1 to 2% by weight of the composition.
7. The composition of claim 1 wherein the K/Na ratio is from about 1/1 to about 3/1.
8. The composition of claim 1 which further comprises an anti-foaming agent.
9. The composition of Claim 1, said composition satisfying the following stability criteria for at least 24 weeks at ambient temperature and for at least 1 week at 140° F: substantially no phase separation, substantially no change in dynamic-mechanical properties, no crystal growth, and no significant color change.
10. A linear viscoelastic aqueous liquid automatic dishwasher detergent comprising, approximately by weight,
  - (a)
    - (i) 8 to 25% potassium triphosphate detergent builder,

- (ii) 2 to 10% sodium tripolyphosphate detergent builder;  
 (b) 5 to 15% alkali metal silicate;  
 (c) 1 to 6% alkali metal hydroxide;  
 (d) 0.1 to 3% chlorine bleach stable, water-dispersible organic detergent active material;  
 5 (e) 0.05 to 1.5% chlorine bleach stable foam depressant;  
 (f) chlorine bleach compound in an amount to provide about 0.2 to 4% of available chlorine;  
 (g) 0.4 to 2% high molecular weight hydrophilic cross-linked polycarboxylate acid thickening agent to provide said linear viscoelastic property;  
 (h) 0.01 to 0.4% of long chain fatty acid or a metal salt of a long chain fatty acid as a physical  
 10 stabilizer to increase the physical stability of the composition;  
 (i) 0 to 10% of a non-crosslinked polyacrylic acid having a molecule weight of from about 800 to 200,000; and  
 (j) water;  
 and wherein in the entire composition the ratio, by weight, of potassium and sodium is from about  
 15 1.05/1 to 3/1, said composition having a bulk density of from about 1.28 g/cc to 1.42 g/cc.

11. The composition of claim 10 which comprises, approximately, by weight,

- (a)  
 (i) 15 to 22% potassium tripolyphosphate;  
 20 (ii) 3 to 8% sodium tripolyphosphate  
 (b) 8 to 12% alkali metal silicate;  
 (c) 1.2 to 4% alkali metal hydroxide;  
 (d) 0.5 to 2% chlorine bleach stable, water-dispersible, low-foaming non-soap anionic detergent  
 active material;  
 25 (e) 0.1 to 0.5% chlorine bleach stable foam depressant;  
 (f) chlorine bleach compound in an amount to provide 0.8 to 1.6% of available chlorine;  
 (g) 0.4 to 1.0% of cross-linked polyacrylic acid having a molecular weight of from about 1,000,000 to  
 4,000,000;  
 (h) 0.08 to 0.3% of stearic acid or a mixture of stearic acid and palmitic acid, or a mixture of fatty  
 30 acids having from 18 to 36 carbon atoms;  
 (i) 0 to 8% of a non-crosslinked polyacrylic acid having a molecular weight of from about 1,000 to  
 150,000;  
 (j) water; and  
 (k) air, in the form of finely divided bubbles, in an amount up to 2% by volume, based on the volume  
 35 of the composition;  
 and wherein the ratio K/Na in said composition is from 1.1/1 to 2.5/1, said composition having a bulk  
 density in the range of from 1.35 g/cc to 1.40 g/cc.

12. The composition of claim 10 wherein (d) comprises alkali metal mono- and/or di- (C<sub>8</sub>-C<sub>11</sub>) alkyl  
 40 diphenyl oxide mono- and/or di-sulphate.

13. The composition of claim 10 in which the chlorine bleach compound (f) is sodium hypochlorite.

14. The composition of claim 10 in which the foam depressant (a) is an alkyl acid phosphate ester, an alkyl  
 45 phosphonic acid ester containing one or two C<sub>12-20</sub> alkyl groups, an ethoxylated product thereof or a  
 mixture thereof.

15. The composition of Claim 11, said composition satisfying the following stability criteria for at least 24  
 50 weeks at ambient temperature and for at least 1 week at 140°F: substantially no phase separation,  
 substantially no change in dynamic-mechanical properties, no crystal growth, and no significant color  
 change.

16. A method for preparing a thickened stable aqueous automatic detergent dishwashing composition  
 which comprises the steps of

- 55 I.  
 (a) fully hydrating a cross-linked high molecular weight polycarboxylate thickener by slowly  
 adding the thickener to water preheated to a temperature of from 95 to 140°F while moderately  
 agitating the mixture,

(b) slowly adding a neutralizing agent comprising caustic soda to the mixture from (a) while continuing agitation to obtain a dispersion of the neutralized thickener;

II.

(c) forming an aqueous mixture of a low-foaming, chlorine-bleach stable, water dispersible surfactant, and optionally one or both of a foam depressant and a fatty acid or fatty acid salt as a stabilizing agent;

(d) heating the mixture in (c) to a temperature higher than 140° F and mixing until a homogeneous smooth surfactant premix emulsion is obtained;

III.

(e) uniformly mixing (i) potassium tripolyphosphate and (ii) sodium tripolyphosphate with the dispersion (b) at an (i)/(ii) ratio from 1.4:1 to 10:1,

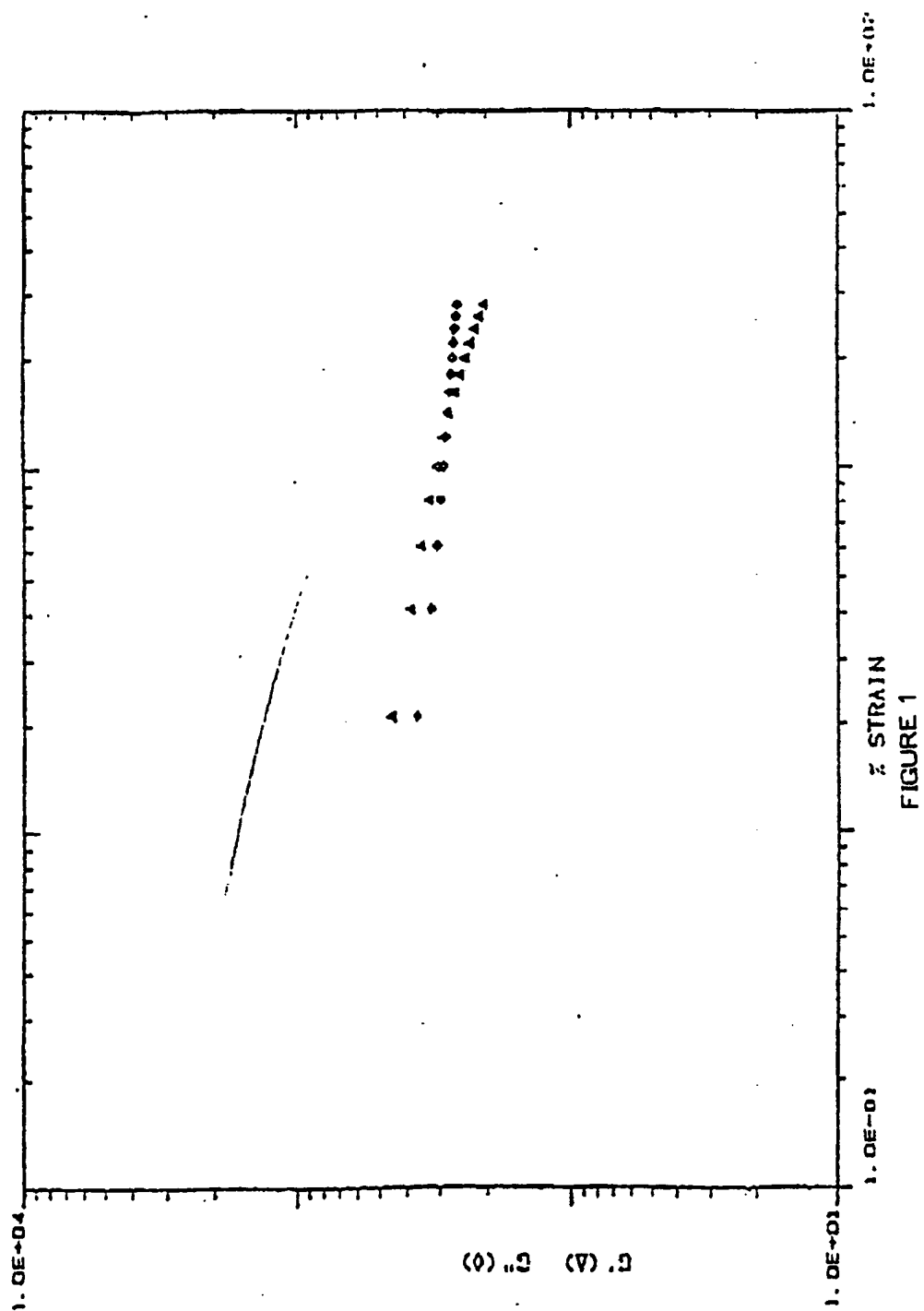
(f) uniformly mixing the heated premix surfactant emulsion (d) with the mixture (e),

(g) cooling the mixture (f) to about 95° F or lower, and

(h) adding bleach to the mixture (g).

17. The method of claim 16 wherein the amount of neutralizing agent is sufficient to raise the pH of the polycarboxylate dispersion to at least 11, and maintaining the pH at each subsequent step to at least 11.

18. The method of claim 17 wherein the pH is 11.5 or more.





2 STRAIN  
FIGURE 2

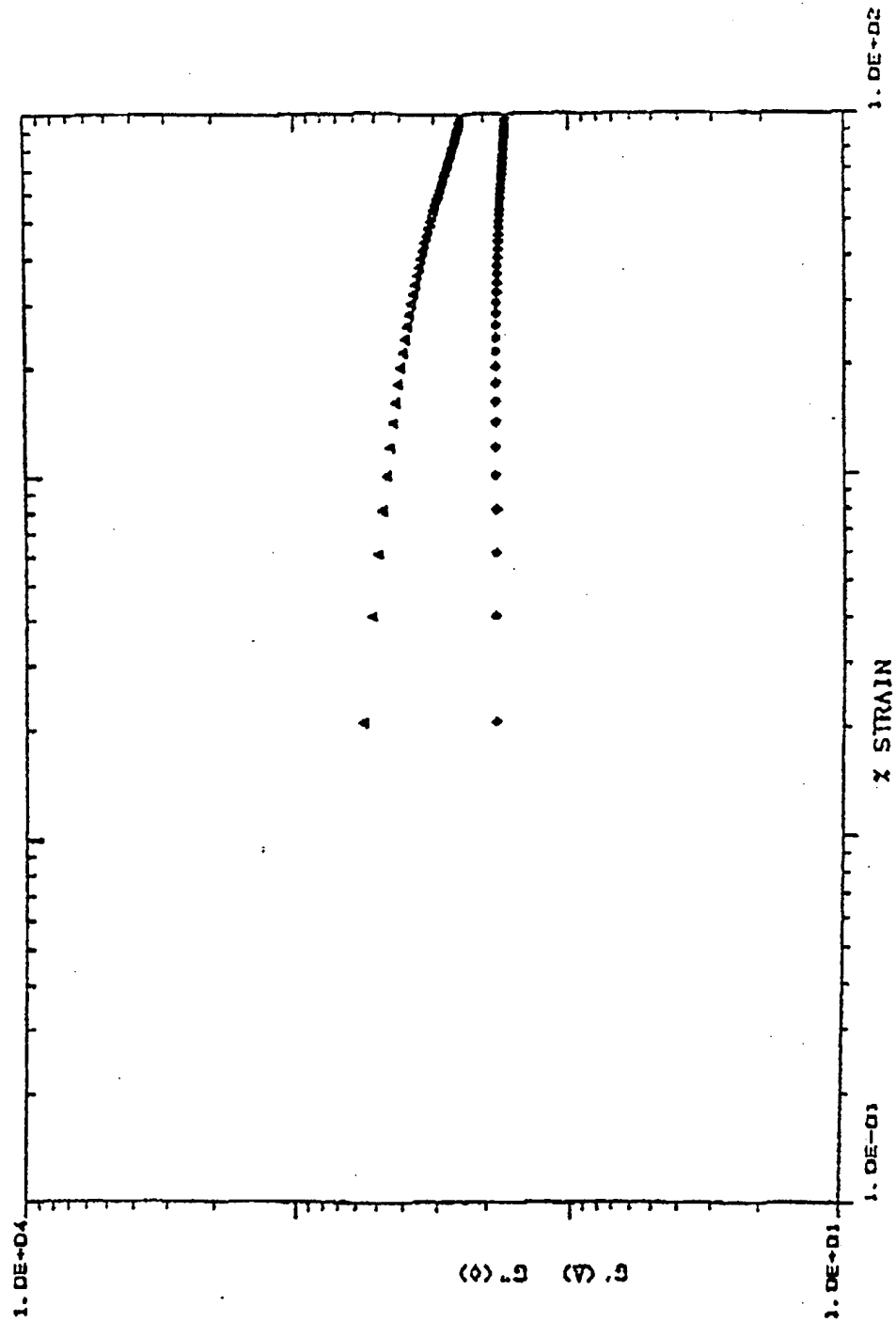
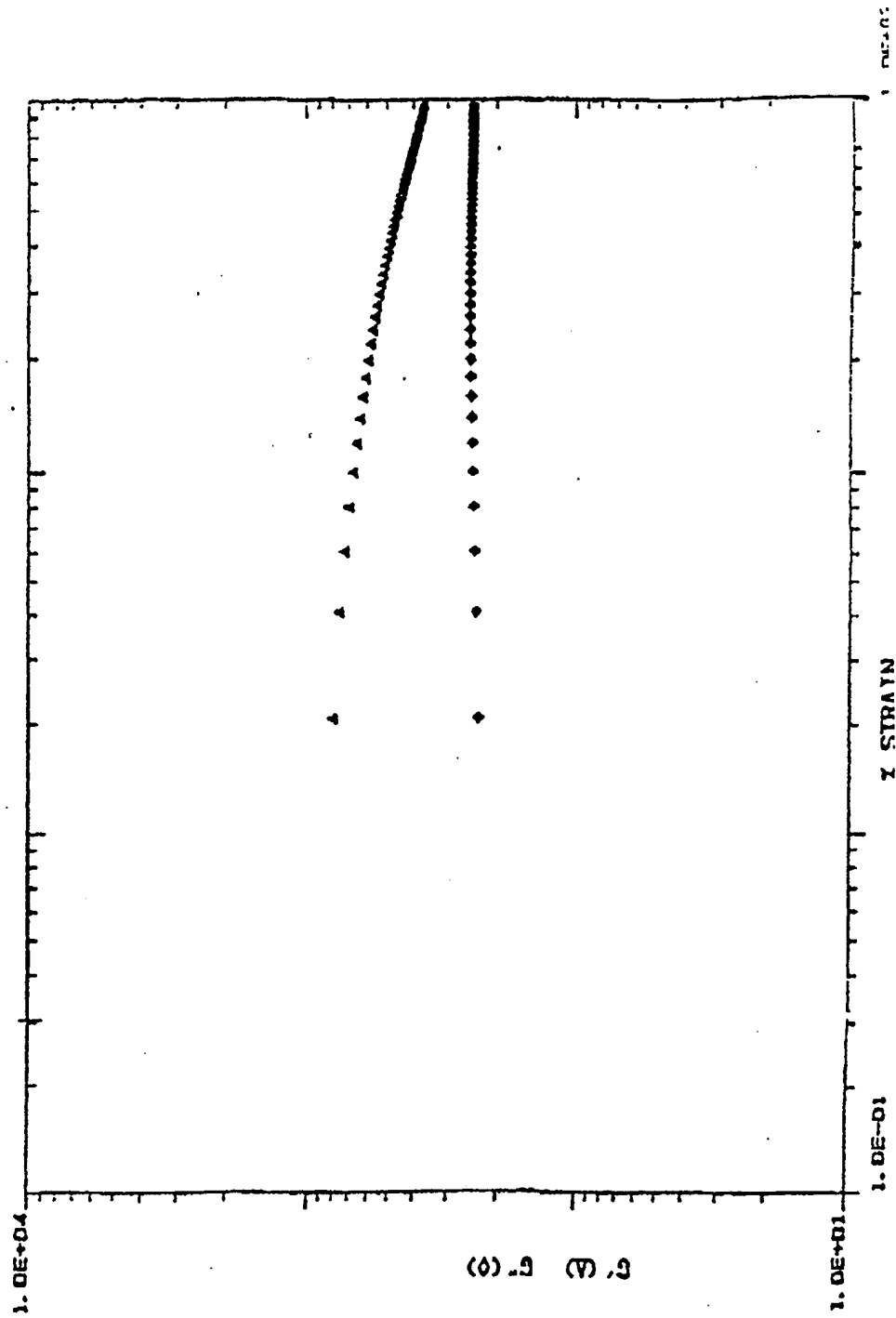


FIGURE 3





Z STRAIN  
FIGURE 4

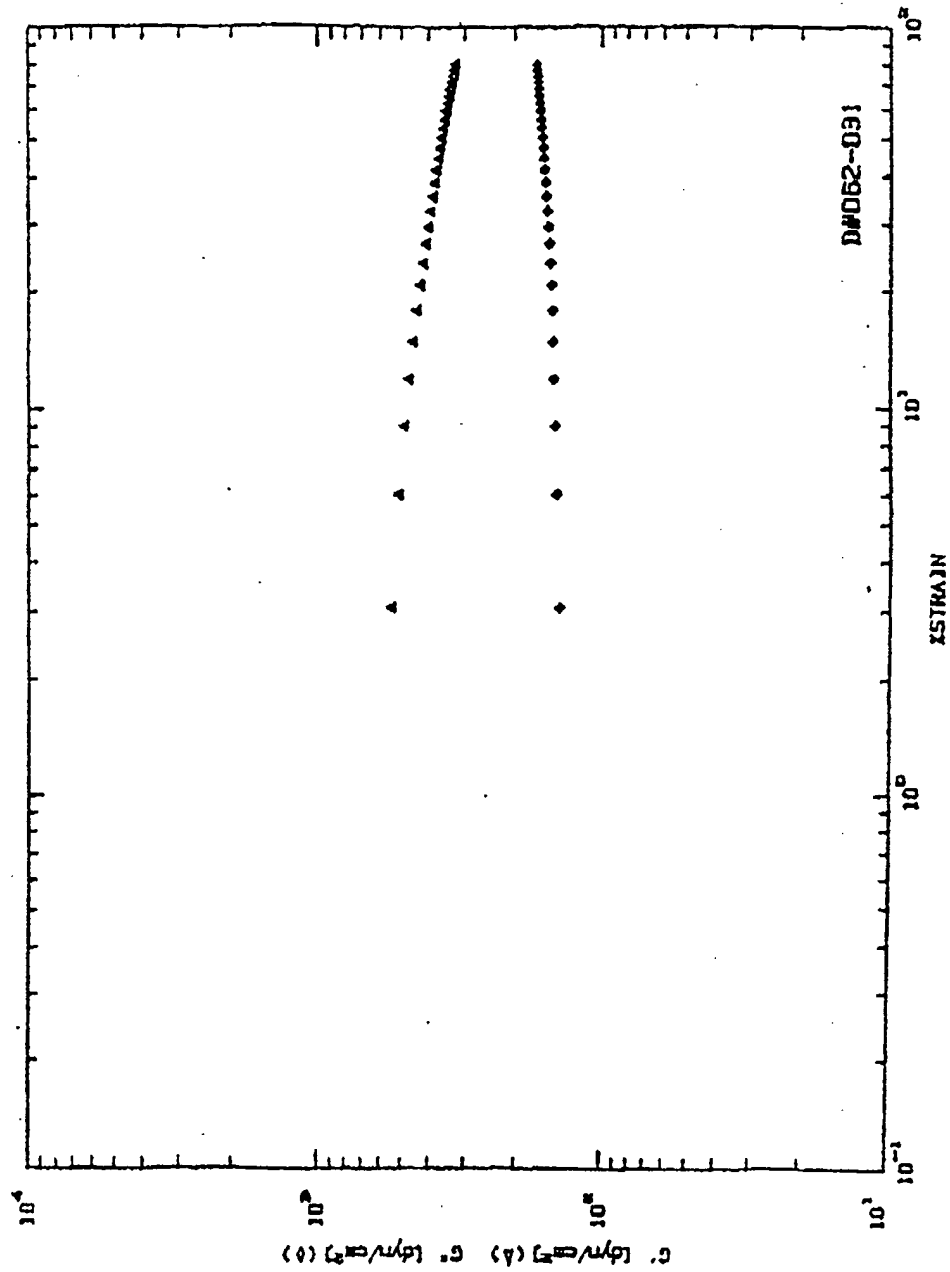
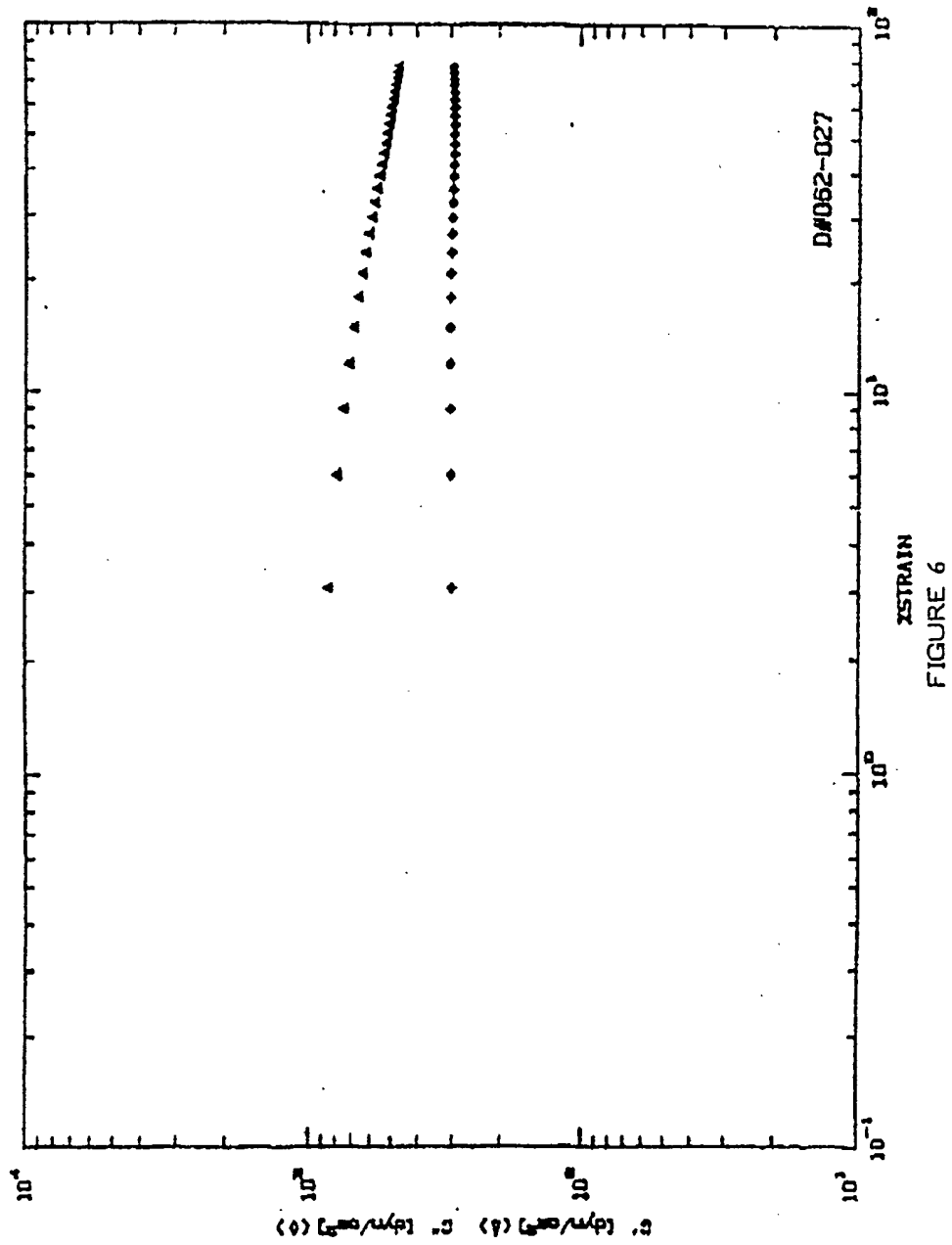


FIGURE 5





European Patent  
Office

# EUROPEAN SEARCH REPORT

Application Number

DOCUMENTS CONSIDERED TO BE RELEVANT			EP 91103284.5
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CL.5)
A	EP - A2 - 0 165 676 (CHEMED CORPORATION) * Examples *	1, 3, 5, 6	C 11 D 3/395 C 11 D 3/06 C 11 D 3/37
A	EP - A1 - 0 264 975 (THE PROCTER & GAMBLE) * Claims 1, 4, 7, 8; page 3, line 30 - page 4, line 19; page 5, line 37 - page 6, line 28 *	1, 3, 5, 6	
P, A	EP - A2 - 0 362 178 (MONSANTO COMPANY) * Example 2 *	1, 3, 5, 6	
D, A	US - A - 4 836 948 (CORRING) * Examples; column 7, lines 13-41 *	1, 5, 6	
D, A	US - A - 4 859 358 (GABRIEL et al.) * Examples II, III; claims 1-6; column 16, lines 37-65 *	1, 2, 5	TECHNICAL FIELDS SEARCHED (Int. CL.5)  C 11 D
The present search report has been drawn up for all claims			
Place of search VIENNA		Date of completion of the search 17-06-1991	Examiner WILFLINGER
<b>CATEGORY OF CITED DOCUMENTS</b> X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document  T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons  A : member of the same patent family, corresponding document			